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STM images of molecularly and atomically chemisorbed oxygen on silver

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Abstract

We present a theoretical study of scanning tunneling microscopy/microscope (STM) images of two chemisorbed, molecular precursors of O₂ and chemisorbed O on Ag(110). The electronic structure and the equilibrium geometric structure of these precursors were obtained from density functional calculations and STM images were calculated from the Kohn–Sham wave functions using the Tersoff–Hamann approximation. This study provides a quantitative analysis of molecular contrast versus molecular orientation, tip–surface distance and character of tip orbital. Our calculated STM images are in good agreement with experimental STM images. We explain the large separation between the experimentally observed protrusions compared to the O–O bond distance using the vacuum tails of a model wave function for an anti-bonding state of O₂.

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Keywords: Scanning tunneling microscopy; Oxygen; Chemisorption; Silver; Density functional theory

1. Introduction

The interaction of oxygen with an (110) surface of silver involves several molecularly adsorbed states and may be viewed as a model system for the study of molecular precursors. The chemisorption properties of this system have attracted a lot of attention both from theoretical and experimental studies [1–9]. In a density functional study by Gravil and Bird two chemisorbed molecular pre-

cursors of oxygen molecules were identified corresponding to two different orientations of the molecule [8]. This prediction was later confirmed by electron energy loss experiments [4]. These precursors were directly imaged in scanning tunneling microscopy/microscope (STM) experiments by Hahn and coworkers [1,2]. They were able to determine the adsorption site of the molecular precursors by imaging the underlying substrate lattice. Using a functionalized tip, they could also resolve protrusions in the images for one of the molecular precursors that were attributed to an anti-bonding molecular π resonance. These protrusions were used to infer the orientation of the molecular precursors. There is a need to under-

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44 stand the nature of these images and what kind of
45 information that can be obtained from them. In
46 particular, when using a functionalized tip, why
47 are protrusions observed in the STM image of only
48 one of the precursors?

49 Several different theoretical schemes for simu-
50 lating STM images of chemisorbed molecules on
51 metal surfaces have been proposed [11,12]. A
52 widely used scheme is based on semi-empirical
53 electronic structure calculations using a non-per-
54 turbative scattering approach to tunneling. An-
55 other scheme that is able to provide a more
56 accurate description of the bonding parameters is
57 based on density functional theory (DFT) and the
58 Tersoff–Hamann approximation for tunneling
59 [13]. The TH approximation gives a simple, phys-
60 ical interpretation in terms of the local density of
61 states (LDOS) at the tip apex and at the Fermi
62 energy. However, the TH approximation is con-
63 troversial because of its simplified treatment of the
64 STM tip, perturbative character and range of ap-
65 plicability for simulating STM images of isolated
66 adsorbates is not clear. Furthermore, the DFT-TH
67 scheme for simulations of STM images have been
68 limited to high coverages and unrealistically short
69 tip–surface distances [14,15] with some contradic-
70 tory results. For example, using the DFT-TH
71 scheme Bocquet et al. [16] were unable to obtain a
72 satisfactory representation of STM images for O₂
73 on Pt(1 1 1) [18], whereas Eichler et al. [17] found a
74 qualitative agreement with the same images using
75 the same scheme. Furthermore, STM image simu-
76 lations of atomic oxygen on metal surfaces using
77 the DFT-TH scheme have reported both a pro-
78 trusion [16] and a depression [15], whereas exper-
79 imental STM images exhibit a depression [12].
80 Thus it is important to explore the usefulness of
81 the DFT-TH scheme by calculating STM images
82 from isolated adsorbates at appropriate tip–sur-
83 face distances and at low coverages. Recently, we
84 have shown that the DFT-TH scheme is able to
85 give a good representation of experimental STM
86 images of adsorbed hydrocarbons on a copper
87 surface, obtained using both regular and func-
88 tionalized tips [10].

89 In this paper we present density functional
90 calculations of chemisorption parameters and
91 STM images of the two molecular precursors of O₂

and of an O atom on Ag(110). The constant 92
current STM images were calculated from the 93
Kohn–Sham wave functions using the TH ap- 94
proximation [13]. The electronic structures of the 95
adsorbates were analyzed using electron density 96
differences and density of states projected on 97
atomic orbitals or linear combinations of atomic 98
orbitals, for O and O₂, respectively. Using the 99
vacuum tails of a model molecular wave function, 100
we discuss the large apparent size of the protru- 101
sions in the STM images. 102

2. Method

103

The density functional calculations of the 104
chemisorbed O₂ molecule and the O atom on 105
Ag(110) were carried out using the plane wave, 106
pseudopotential code DACAPO [21]. The interac- 107
tion of the valence electrons with the ion cores 108
were treated using ultra-soft pseudopotentials [22] 109
with a plane-wave energy cutoff of 35 Ry and the 110
exchange and correlation effects were included 111
using the generalized gradient approximation [23]. 112
The calculation of STM images requires a large 113
surface unit cell so the surface was modeled by a 114
slab in supercell geometry with a (4 × 3) surface 115
unit cell and six layers of Ag atoms and a vacuum 116
region being equivalent to six layers. A smaller 117
(3 × 2) surface unit cell was used in the analysis of 118
the electronic structure and in the calculation of 119
the potential energy barrier between the two ori- 120
entations of the O₂ molecules. This barrier was 121
calculated by carrying out a constrained relaxation 122
of the atoms in a series of calculations where the 123
O–O axis is constrained to lie in a plane perpen- 124
dicular to the surface with different angles with 125
respect to a symmetry plane of the surface. We 126
used 16 *k*-points in the surface Brillouin zones 127
(SBZ) of the small and large super cells. The 128
equilibrium geometry were determined by relaxing 129
the O positions and the three outermost Ag layers 130
whereas the remaining Ag layers were fixed at their 131
calculated bulk positions. 132

The STM images were calculated using the 133
Tersoff and Hamann (TH) approximation for 134
tunneling in an STM junction [13]. At zero tem- 135
perature and low bias voltage, *V*, the differential 136

137 tunneling conductance, dI/dV is in this approxi-
 138 mation proportional to the LDOS at the tip apex,
 139 \mathbf{r}_0 , and at the Fermi energy, ϵ_F , which is given by,

$$\rho(\mathbf{r}_0; \epsilon_F) = \sum_{n\mathbf{k}} |\psi_{n\mathbf{k}}(\mathbf{r}_0)|^2 \delta(\epsilon_{n\mathbf{k}} - \epsilon_F). \quad (1)$$

141 Here $\psi_{n\mathbf{k}}(\mathbf{r})$ is a Kohn–Sham wave function of the
 142 adsorbate-covered surface with energy $\epsilon_{n\mathbf{k}}$, band
 143 index n and wave vector \mathbf{k} in the SBZ. The TH
 144 result for dI/dV is based on the s-wave approxi-
 145 mation for the tip. For a p_z -wave approximation
 146 for the tip, dI/dV is proportional to an analogous
 147 quantity to the LDOS in Eq. (1) obtained by re-
 148 placing $\psi_{n\mathbf{k}}(\mathbf{r}_0)$ with $\partial\psi_{n\mathbf{k}}(\mathbf{r}_0)/\partial z$ [19]. The constant-
 149 current STM images and profiles correspond to
 150 topographical images and profiles of constant
 151 $\rho(\mathbf{r}_0; \epsilon_F)$, given by Eq. (1), and will be referred to as
 152 LDOS images and profiles, respectively. Before
 153 presenting and discussing these results, we need
 154 first to present and discuss the results for the
 155 geometric and electronic structure of the adsor-
 156 bates.

157 3. Chemisorption parameters

158 We have considered the same configurations for
 159 the two molecular precursors and the atom as
 160 done in the DFT investigation by Gravil et al. [8].
 161 Our results for bonding parameters are in good
 162 agreement with their results considering the use of
 163 different pseudopotentials. In particular, the use of
 164 ultra-soft pseudopotentials for oxygen is delicate
 165 with special needs in handling the core electrons
 166 [24]. The two molecular configurations correspond
 167 to the O_2 molecule being adsorbed in the hollow
 168 site either with the O–O axis in the $[00\bar{1}]$ or in the
 169 $[1\bar{1}0]$ directions, as shown in Fig. 1(a) and (b),
 170 respectively. These two molecular precursors will
 171 henceforth be referred to as $O_{2[00\bar{1}]}$ and $O_{2[1\bar{1}0]}$,
 172 respectively. The O atom is adsorbed in the four-
 173 fold hollow site, as shown in Fig. 1(c).

174 The bonding distances for the two molecular
 175 orientations are somewhat different but the ener-
 176 getics are very similar. The O–O bond distance is
 177 1.39 \AA for $O_{2[00\bar{1}]}$ which is 0.03 \AA smaller than for
 178 $O_{2[1\bar{1}0]}$. The equilibrium height of the molecule
 179 above the surface plane is about 1.33 \AA for $O_{2[00\bar{1}]}$,

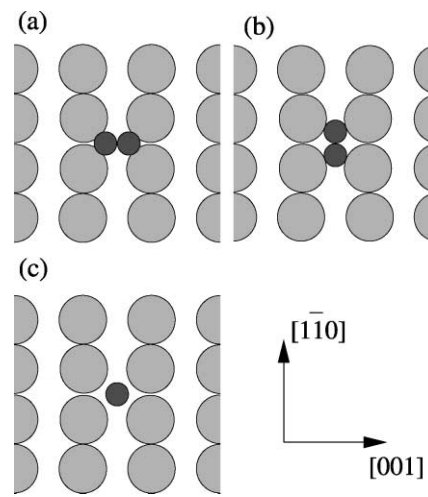


Fig. 1. Surface unit cell and chemisorption geometries for O_2 (a,b) and O (c) on Ag(110). The molecular axis of O_2 is along the $[00\bar{1}]$ direction ($O_{2[00\bar{1}]}$) in (a) and along the $[1\bar{1}0]$ direction ($O_{2[1\bar{1}0]}$) in (b). Large and small circles represent Ag and O atoms, respectively.

180 which is 0.19 \AA higher than for $O_{2[1\bar{1}0]}$. This dif-
 181 ference in adsorption height is simply a geometri-
 182 cal effect of keeping the O–Ag bond distance more
 183 or less the same for the two molecular precursors.
 184 Our calculated chemisorption energy is 0.37 eV for
 185 both orientations, which is consistent with the
 186 experimental observation that both precursors are
 187 observed at the same surface temperature. We
 188 found that the potential energy barrier between the
 189 two orientations is 0.45 eV and is located at an
 190 angle of 45° for the O–O axis with respect to the
 191 $[00\bar{1}]$ direction. This result suggest that the mole-
 192 cule will rather desorb than switch orientation
 193 when elevating the temperature. Note that Gravil
 194 and Bird only estimated the location of this barrier
 195 and hence its magnitude.

196 As expected for an open shell atom, the O atom
 197 forms a strong chemisorption bond with a metal
 198 surface such as Ag(110). In the fourfold hollow
 199 site, the atom-surface distance is 0.57 \AA and the
 200 adsorption energy is 3.50 eV .

201 The nature of the electronic states of the
 202 chemisorbed molecular precursors and atom is
 203 revealed by the PDOS on the bonding and anti-
 204 bonding π orbitals of O_2 and the PDOS on the p

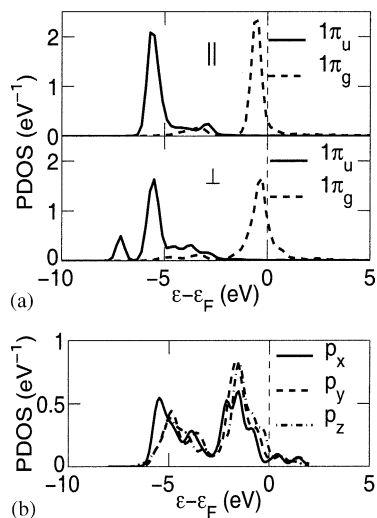


Fig. 2. Calculated projected density of states (PDOS) for $O_{2[110]}$ (a) and O (b), chemisorbed on Ag(110). (a) The density of states are projected on linear combination of atomic p orbitals modeling the bonding and anti-bonding π molecular orbitals of O_2 oriented parallel to the surface in top panel ($1\pi_{\parallel}^{\parallel}$ and $1\pi_{\parallel}^{\perp}$) or perpendicular to the surface in lower panel ($1\pi_{\perp}^{\parallel}$ and $1\pi_{\perp}^{\perp}$). (b) The density of states are projected on atomic p orbitals.

205 orbitals of O, respectively. The PDOS for $O_{2[110]}$
 206 and $O_{2[001]}$ are very similar and only the PDOS for
 207 $O_{2[110]}$ is shown in Fig 2(a). Around the Fermi
 208 level and above the energy range of the d band, the
 209 PDOS is dominated by narrow molecular reso-
 210 nances derived from anti-bonding molecular states
 211 $1\pi_{\parallel}^{\parallel}$. The resonance that is derived from the $1\pi_{\parallel}^{\parallel}$
 212 state where the atomic p states are parallel to the
 213 surface is almost filled, whereas the resonance that
 214 derives from the $1\pi_{\parallel}^{\perp}$ where the atomic p states are
 215 perpendicular to the surface is partially filled. The
 216 net donation of more than one electron to the anti-
 217 bonding π orbitals of the O_2 molecule results in a
 218 weakening of the intra-molecular bond and an
 219 elongation of its bond distance with about 0.15 Å
 220 upon adsorption. As shown in Fig. 2(b), the p
 221 subshell of the O atom hybridizes strongly with the
 222 d band and is almost completely filled upon
 223 chemisorption, indicating an O^{2-} charged state.

224 The filling and the characteristic spatial depen-
 225 dence of the anti-bonding π states of O_2 and the p

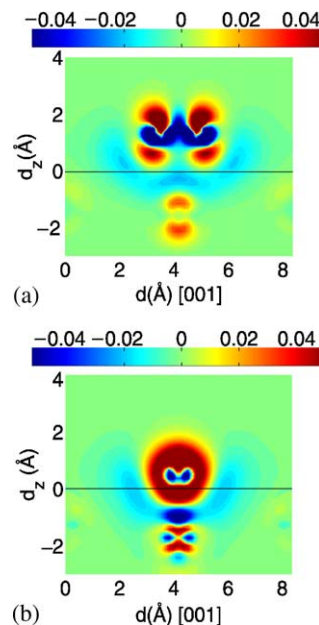


Fig. 3. Density difference contours of (a) $O_{2[001]}$ and (b) O on Ag(110). The contour planes are perpendicular to the surface (d_z) and along the [001] direction, through the adsorbate. Density differences for values larger than 0.05 and less than -0.05 electrons/Å³ have been truncated. $d_z = 0$ corresponds to the position of the first atomic surface layer. The calculations were carried out in a 3×2 surface unit cell with 16 k -points in the SBZ.

226 states of O upon chemisorption is directly revealed
 227 in the contours of electron density differences,
 228 $\delta n(\vec{r})$, in Fig. 3. In both cases, the contours of $\delta n(\vec{r})$
 229 show also that the d states of the nearest neigh-
 230 boring atom in the second layer of Ag are partic-
 231 ipating in the chemisorption bond. Another
 232 common feature that is present in the density dif-
 233 ference contours for both O_2 and O is the region of
 234 electron density depletion in the vicinity of the
 235 adsorbate. This depletion is due to a decrease in
 236 the free-electron like density of metal states and is
 237 associated with the two related effects: (1) electron
 238 transfer from the free-electron like states into the π
 239 or p states of O_2 and O, respectively, (2) screening
 240 of the negative charge transferred into the adsorb-
 241 ate states by the free-electron like metal states.

242 4. Simulated STM images

243 Based on our results for the geometric and
244 electronic structure we can now make a compre-
245 hensible presentation and discussion of the results
246 for the LDOS images and profiles of $O_{2[1\bar{1}0]}$ and
247 $O_{2[001]}$ and the chemisorbed O atom shown in Figs.
248 4–6, respectively. A common feature of all the
249 LDOS images and profiles is the wide depression,
250 which extends about 4–5 Å from the center of the
251 adsorbate. A characteristic feature of the molecu-
252 lar LDOS images and profiles are the two pro-
253 trusions along the molecular axis of O_2 , which are
254 much more pronounced for $O_{2[1\bar{1}0]}$ than for $O_{2[001]}$.
255 These two features are now discussed in turn.

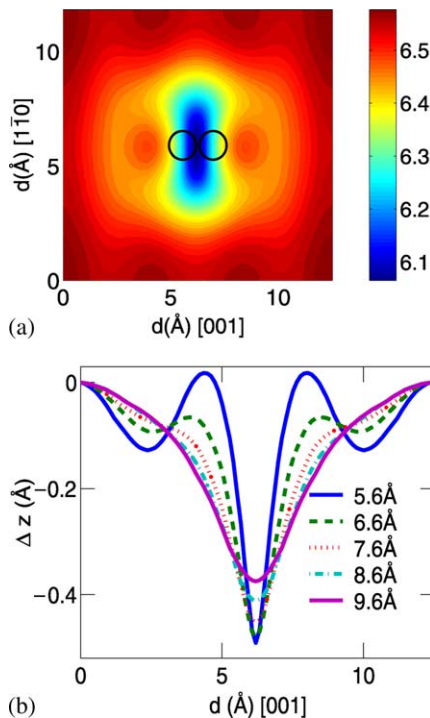


Fig. 4. Calculated LDOS (a) images and (b) profiles along the molecular axis for $O_{2[001]}$ on $Ag(110)$. $\Delta z = 0$ corresponds to an initial tip–surface distance of $z_0 = 5.6, 6.6, 7.6, 8.6,$ and 9.6 Å for the profiles and to 6.6 Å for the image. The geometric configuration of $O_{2[001]}$ is indicated by the two black circles in the LDOS image. The discrete spectrum of Kohn–Sham states was broadened by a Gaussian approximation to the delta function in Eq. (1) using a width of $\sigma = 0.25$ eV. The calculations were carried out in a 4×3 surface unit cell with 16 k -points in the SBZ.

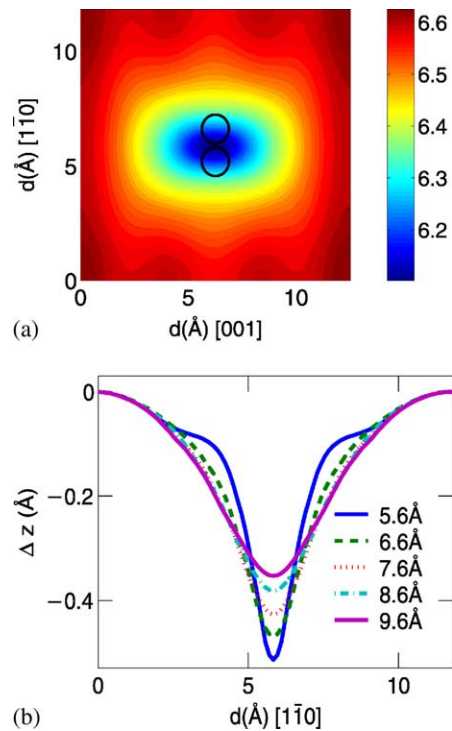


Fig. 5. Calculated LDOS (a) images and (b) profiles along the molecular axis for $O_{2[1\bar{1}0]}$ on $Ag(110)$. $\Delta z = 0$ corresponds to an initial tip–surface distance of $z_0 = 5.6, 6.6, 7.6, 8.6,$ and 9.6 Å for the profiles and to 6.6 Å for the image. The configuration of $O_{2[1\bar{1}0]}$ is indicated by the two black circles in the LDOS image. Same supercell, k -points and σ , as in Fig. 4.

The wide depression in the LDOS images is associated with the decrease of metal free-electron like density in the vicinity of the adsorbate, due to screening of the electronegative adsorbate. This interpretation is corroborated by two findings: (1) the size of the depression in the LDOS images are about the same as the electron depletion region of the electron density difference, as shown in Fig. 3, (2) the development of the depression with increasing tip–surface distances is moderate which shows that these states has a decay into the vacuum which is similar to that of the clean $Ag(110)$, that is, mainly free-electron like.

The two protrusions in the LDOS images of $O_{2[1\bar{1}0]}$ and $O_{2[001]}$ are in both cases associated with the $1\pi_g^\perp$ derived resonance that has a large contribution at the Fermi level and two perpendicular lobes. In the LDOS image these lobes will give rise

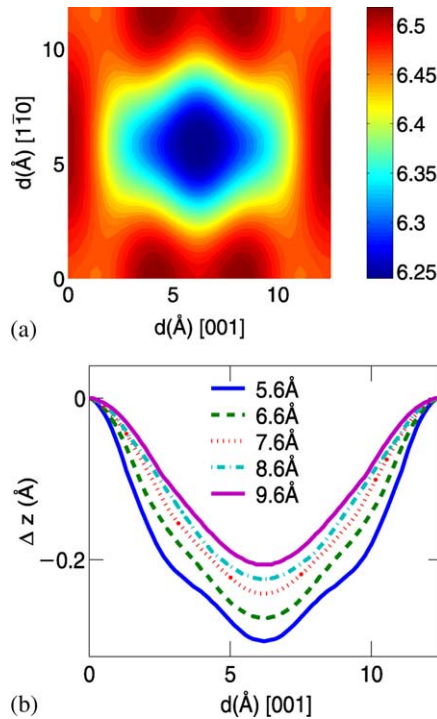


Fig. 6. Calculated LDOS (a) images and (b) profiles along the [001] direction of O on Ag(110). $\Delta z = 0$ corresponds to an initial tip–surface distance of $z_0 = 5.6, 6.6, 7.6, 8.6, 9.6$ Å for the profiles and 6.6 Å for the image. The O is located in the center of the LDOS image. Same supercell, k -points and σ , as in Fig. 4.

274 to two protrusions with a relatively large separation
275 tion, which can be understood using a simple
276 model for the contribution of the $1\pi_g^\perp$ derived
277 resonance at the tip apex. This contribution is
278 modeled by the density $n_{1\pi_g^\perp}(\mathbf{r}_0)$ of a model wave
279 function obtained from an anti-symmetric combi-
280 nation of the vacuum tails of two states with p_z
281 symmetry as,

$$n_{1\pi_g^\perp}(\mathbf{r}_0) = \frac{1}{2} \left| \frac{\partial}{\partial z} \psi_s(\mathbf{r}_0 - \mathbf{r}_1) - \frac{\partial}{\partial z} \psi_s(\mathbf{r}_0 - \mathbf{r}_2) \right|^2 \quad (2)$$

283 where \mathbf{r}_0 is the position of the tip apex, $\mathbf{r}_{1,2}$ are the
284 positions of the two O atoms, z is the direction
285 perpendicular to molecular defined by the p_z
286 atomic orbitals and $\psi_s(\mathbf{r}) \propto e^{-\kappa r}/\kappa r$ [19], where the
287 decay constant $\kappa = \sqrt{2m_e\phi}/\hbar \approx 1.1 \text{ \AA}^{-1}$ is deter-
288 mined by the calculated work function ϕ of about
289 4.3 eV. The profiles of $n_{1\pi_g^\perp}(\mathbf{r}_0)$, shown in Fig. 7,

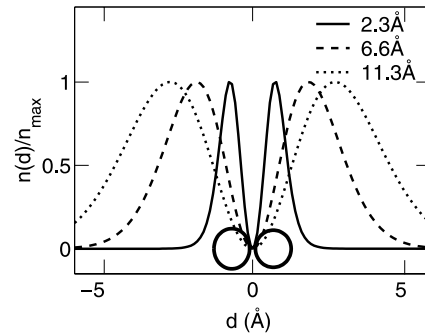


Fig. 7. Calculated densities of a single $1\pi_g^\perp$ model wave function along a line through the molecular axis at different tip–surface distances of $z_0 = 2.3, 6.6,$ and 11.3 Å. The positions of the O atoms are indicated by the open circles. The densities, $n(d)$, are constructed using Eq. (2), and are normalized to their maximum value, n_{\max} , along the line.

have two protrusions separated by distance, d_p , 290
that varies strongly with the tip–surface distance, 291
 z_0 . At close distance it converges to O–O bond 292
distance, $d_{\text{O–O}}$, of about 1.4 Å whereas at $z_0 = 6.6$ 293
Å, $d_p = 3.7$ Å, which is comparable to $d_p = 4.6$ Å 294
in the LDOS image at the same tip–surface distance, 295
shown in Fig. 4. For large tip–molecule 296
distances, $z_m, \kappa z_m \gg 1$ and $z_m/d_{\text{O–O}} \gg 1$, the varia- 297
tion of d_p with z_m has a simple analytic form, 298
 $d_p \sim 2\sqrt{z_m/\kappa}$, that can be obtained from the asymp- 299
totic behavior of $n_{1\pi_g^\perp}(\mathbf{r}_0) \sim |(\partial^2/\partial z \partial x)\psi_s \times$ 300
 $(\mathbf{r}_0 - (\mathbf{r}_1 + \mathbf{r}_2)/2)|^2$, where x is the coordinate 301
along the molecular axis. For instance, at $z_m = 5.3$ 302
Å, corresponding to $z_0 = 6.6$ Å, $2\sqrt{z_m/\kappa} \sim 4.4$ Å. 303

There are primarily two reasons why the contribu- 304
tion from the $1\pi_g^\perp$ resonance dominates over 305
the contribution from the $1\pi_g^\parallel$ contribution to the 306
LDOS image. First, the PDOS in the former state 307
is larger at the Fermi energy than the latter state, 308
as shown in Fig. 2. Second, the wave function of 309
the $1\pi_g^\perp$ gives a relatively larger contribution in the 310
vacuum region than the wave function of the $1\pi_g^\parallel$. 311
This effect can be shown directly from the asymp- 312
totic behavior of the model wave functions of 313
these states, using Eq. (2) for $1\pi_g^\perp$ and a similar 314
construction using p_y model wave functions for 315
 $1\pi_g^\parallel$, see Ref. [19]. For large κz_m and $z_m/d_{\text{O–O}}$ the 316
ratio between the maximum value of the model 317
densities is given by, 318

$$\frac{n_{1\pi_g^{\parallel}}}{n_{1\pi_g^{\perp}}} = \left(\frac{y}{z_m}\right)^2 \quad (3)$$

320 The maximum value of $n_{1\pi_g^{\perp}}(\mathbf{r}_0)$ occurs for
 321 $y \sim \sqrt{2z_m/\kappa}$ and its contribution to the LDOS is
 322 $2/\kappa z_m$ relative to $n_{1\pi_g^{\parallel}}(\mathbf{r}_0)$. At $z_m = 5.3$ and 10 \AA ,
 323 this factor is about 3 and 5, respectively.

324 We argue that the more pronounced protrusion
 325 in the LDOS image for $O_{2[001]}$ than for
 326 $O_{2[1\bar{1}0]}$ is simply a consequence of the differences in
 327 the adsorption height. The difference in the
 328 strengths of the protrusions is not simply a consequence
 329 of differences in the interaction of the
 330 $1\pi_g^{\perp}$ state with the metal states because of two
 331 findings: (1) the corresponding PDOS for the two
 332 molecular precursors are almost indistinguishable
 333 around the Fermi level (2) the calculated adsorption
 334 energy, which is a measure of the interaction
 335 strength, is the same for the two configurations.
 336 However, the closer distance of $O_{2[1\bar{1}0]}$ to the surface
 337 than $O_{2[001]}$ suggests that the contribution
 338 from the resonance state derived from the $1\pi_g^{\perp}$
 339 state to the LDOS is smaller relative to the contribution
 340 from metal states for $O_{2[1\bar{1}0]}$ than for
 341 $O_{2[001]}$. Note that the correspondence between the
 342 corrugation of an LDOS image and the molecule–
 343 surface distance is non-linear. The corrugations of
 344 the LDOS profiles at 5.6 \AA for $O_{2[1\bar{1}0]}$ and at 7.6 \AA
 345 for $O_{2[001]}$ are similar as shown in Figs. 4 and 5.
 346 Thus, to keep the same corrugation of the LDOS
 347 image for an outward shift of the molecule of 0.2
 348 \AA we need to increase z_0 by about 2 \AA at 5.6 \AA .

349 In our calculated LDOS images of adsorbed O
 350 we find only a wide depression and no protrusion,
 351 as shown in Fig. 6. The absence of any protrusion
 352 in this case is understood from the electronic structure
 353 and adsorption distance of O on
 354 Ag(110). The contributions of the resonances
 355 derived from the p subshell of states of the O atom
 356 to the PDOS at the Fermi level are small. Furthermore,
 357 the contribution to LDOS from tunneling through
 358 metal-like states is favored over the
 359 contribution from tunneling through adsorbate-
 360 like states by the close adsorption distance of the O
 361 atom to the first layer of substrate atoms.

362 Our calculated LDOS images for $O_{2[001]}$ and
 363 $O_{2[1\bar{1}0]}$ are in near-quantitative agreement with the

experimental STM images by Hahn and coworkers 364
 [1,2]. Using a functionalized tip that was formed 365
 by a transfer of a single CO molecule from the 366
 surface to the tip, they were able to enhance the 367
 resolution of the STM images of the molecular 368
 precursors and to resolve two protrusions in the 369
 STM image for one of the molecular precursors. 370
 The observed spatial extension and depth of the 371
 depressions in the images for two molecular pre- 372
 cursors and the observed height and distance be- 373
 tween the two protrusions in the image $O_{2[001]}$ are 374
 nicely reproduced by our calculated LDOS images 375
 at $z_0 \approx 7 \text{ \AA}$. For STM images obtained with a 376
 regular tip the experimental STM images of the 377
 two molecular precursors and chemisorbed O are 378
 in good agreement with the calculated LDOS im- 379
 ages at $z_0 \approx 9 \text{ \AA}$. Also Zambelli et al. [6] have 380
 measured STM images of O_2 chemisorbed on 381
 Ag(110), where O_2 is imaged as a symmetric de- 382
 pression similar to the regular tip STM images of 383
 $O_{2[1\bar{1}0]}$ and $O_{2[001]}$ by Hahn and coworkers. Note 384
 that at large tip–surface distances only a large 385
 uncharacteristic depression appear in the LDOS 386
 images and profiles for the chemisorbed molecule 387
 and atom. In particular, this makes it difficult to 388
 discriminate between their chemical identities from 389
 STM images obtained using a regular tip. Fur- 390
 thermore, the elongation of the depression is per- 391
 pendicular to the molecular axis, which shows that 392
 the orientation of the molecule cannot be inferred 393
 simply from the shape of its STM image. 394

395 We have simulated STM images using a p_z ap-
 396 proximation of the tip. One could argue that this
 397 model should represent better a CO-functionalized
 398 tip considering the electronic structure of a CO
 399 molecule, where the CO 5σ orbital is expected to
 400 give the main contribution [16]. However, we find
 401 that the simulated STM images using the p_z -wave
 402 approximation gives essentially the same images as
 403 the s-wave approximation at a tip–surface distance
 404 that is about 1 \AA larger than for the s-wave ap-
 405 proximation, as shown in Fig. 8.

5. Summary

406

We have carried out density functional calcula- 407
 tions of STM images of two molecular precur- 408

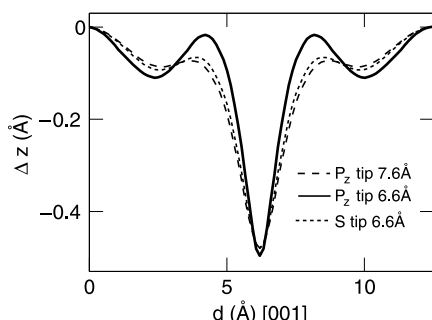


Fig. 8. Simulated STM profiles for $O_{2[001]}$ along the $[001]$ using an s-wave tip (dashed line), corresponding to LDOS image of Fig. 4, and a p_z -wave tip (solid line). $\Delta z = 0$ corresponds to $z_0 = 6.6$ or 7.6 Å, as indicated in the figure. Same supercell, k -points and σ , as in Fig. 4.

409 sors of O_2 and O chemisorbed on Ag(110) using
 410 the TH approximation for tunneling. We find a
 411 good agreement between experimental STM im-
 412 ages of the molecular precursors obtained using
 413 both regular and functionalized tips and chemi-
 414 sorbed O obtained using a regular tip and our
 415 LDOS images evaluated at appropriate tip–surface
 416 distances. This shows that density functional cal-
 417 culations together with the TH approximation is a
 418 useful scheme for simulating STM images of single
 419 adsorbates. At large tip–surface distances the
 420 LDOS images are similar so that STM images
 421 from regular tips do not simply discriminate be-
 422 tween chemisorbed O and O_2 . Our analysis of the
 423 calculated LDOS images shows that the observed
 424 protrusions derive from tunneling through a single
 425 molecular resonance state of $O_{2[001]}$ introduced by
 426 the $1\pi_g^\perp$ molecular orbital of O_2 . The strengths of
 427 these protrusions are drastically reduced for the
 428 $O_{2[1\bar{1}0]}$ compared to $O_{2[001]}$ because the shorter
 429 molecule–surface distance for $O_{2[1\bar{1}0]}$ than for
 430 $O_{2[001]}$ decreases the tunneling through the mole-
 431 cule relative to the substrate. We show that the
 432 large apparent separation between the two pro-
 433 trusions compared to the O–O bond distance is
 434 caused by the behavior of the vacuum tails of the
 435 wave functions associated with the resonance de-
 436 rived from $1\pi_g^\perp$ state.

6. Uncited reference

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